Online Gas Analysis in Ammonia Plants

Process Overview
Ammonia is used in the production of a variety of products including fertilizer, nitric acid, nylon, pharmaceuticals, and refrigerant.

Production of ammonia (NH₃) is a two step process. The first step is to produce hydrogen (H₂) by reacting natural gas or methane (CH₄) with steam in a reformer. The second step is to react hydrogen with nitrogen (N₂) in a 3 : 1 ratio over a high-temperature catalyst in the ammonia synthesis process.

In the process of reforming natural gas or methane to hydrogen, carbon monoxide (CO) and carbon dioxide (CO₂) are formed as by-products. Even at low concentrations, both CO and CO₂ cause irreversible problems in the ammonia synthesis process. When the synthesis gas contains CO, a slow permanent deterioration of the catalyst occurs. The presence of CO₂ in the synthesis gas can result in the formation of ammonium carbamate, a fine white powder, which precipitates out and causes plugging of the ammonia converter. Therefore, to prevent problems in the ammonia synthesis process and to prolong the life of the expensive catalyst, the removal and measurement of CO and CO₂ on a continuous basis is required.

Gas Analyzer Applications
During the production of ammonia, the gases pass through the steam reformer, high- and low-temperature shift converters, amine scrubber and methanator before entering the ammonia synthesis process.

![Figure 1 - Gas Analyzer Applications](image-url)
Measuring Point 1: Steam Reformer

Natural gas and steam react over catalyst to form H₂ and CO:

\[ \text{CH}_4 + \text{H}_2\text{O} \rightarrow 3\text{H}_2 + \text{CO} \]

Also present is CO₂ and unreacted CH₄. Reformer efficiency is monitored by measuring unconverted CH₄. CO is also measured in preparation for the shift converters.

Measuring Points 2A and 2B: High - and Low-Temperature

The shift converters remove CO by reacting with steam to form H₂ and CO₂:

\[ \text{CO} + \text{H}_2\text{O} \rightarrow \text{H}_2 + \text{CO}_2 \]

The CO content is measured to determine the efficiency of the shift converters.

Measuring Point 3: Amine Scrubber

CO₂ is absorbed in amine scrubber. The CO₂ is measured to determine scrubber efficiency.

Measuring Point 4: Methanator

The Methanator removes the remaining traces of CO and CO₂ by converting them to methane:

\[ \text{CO} + 3\text{H}_2 \rightarrow \text{CH}_4 + \text{H}_2\text{O} \]
\[ \text{CO}_2 + 4\text{H}_2 \rightarrow \text{CH}_4 + 2\text{H}_2\text{O} \]

Trace CO and CO₂ are measured because they must be removed before the ammonia synthesis process or the CO will poison the ammonia converter catalyst and the CO₂ will react with ammonia and cause plugging of the ammonia converter. H₂ and CH₄ are also measured to control the feed for ammonia synthesis (described in next section).

The X-STREAM analyzer is extremely well-suited to make the CH₄, CO and CO₂ measurements described above using NDIR (non-dispersive infrared) photometric detectors. Typical analysis ranges after the Methanator are 0 to 10 ppm CO and 0 to 5 ppm CO₂. Abnormally high concentration levels should be alarmed.

Ammonia Synthesis Process

Hydrogen - nitrogen rich gas from the Methanator goes into the conversion - separation loop where it is mixed with recycle gas from the Ammonia Converter. In the Ammonia Converter hydrogen reacts with nitrogen over catalyst to make ammonia.